

## 4',5a'-Diphenyl-10-oxospiro[phenanthrene-[9,2']oxeto[5,4-b]oxazole]

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## Key indicators

Single-crystal X-ray study

$T = 293\text{ K}$

Mean  $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$

$R$  factor = 0.042

$wR$  factor = 0.117

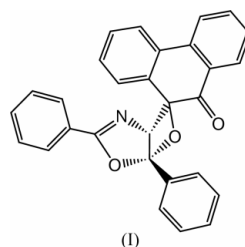
Data-to-parameter ratio = 11.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound,  $\text{C}_{29}\text{H}_{19}\text{NO}_3$ , the phenanthrene is out of planarity with its dihydrobenzene ring being distorted towards a half-chair conformation. The relative configuration of the planar oxetane and oxazole rings is conditioned by the  $\text{C}sp^3$  state of the shared atoms.

## Comment

In our ongoing research on the photocycloaddition reactions of  $\alpha$ -dicarbonyl compounds with oxazole derivatives (Zhang *et al.*, 2003), we have carried out the photoreaction of 9,10-phenanthrene-9,10-dione with 2,5-diphenyloxazole and isolated the title compound, (I). We have also structurally analysed (I) and the results are presented here.



The bond lengths and angles in (I) (Fig. 1) are within normal ranges (Allen *et al.*, 1987). Those within the fused four/five-membered rings ( $\text{C}1/\text{C}16/\text{O}2/\text{C}3/\text{O}1/\text{C}2/\text{N}1$ ) are comparable to the corresponding values in related structures, *viz.* spiro[indole-3,2'-(2'*aH*,5'*aH*)oxeto[5,4-*b*]oxazol]-2(1*H*)-one derivatives (Usman *et al.*, 2003). As in these structures, there is elongation of the  $\text{C}1-\text{C}16$  bond and planarity of the oxetane ring. The dihedral angle between the  $\text{O}2/\text{C}3/\text{C}1$  and  $\text{O}2/\text{C}1/\text{C}16$  planes is  $177.0(1)^\circ$  [*cf.*  $176.9(3)$ – $178.6(2)^\circ$ ; Usman *et al.*, 2003].

The planes through the oxetane and oxazole rings make a dihedral angle of  $113.4(1)^\circ$ . This is a consequence of the  $sp^3$  hybridization of the shared atoms  $\text{C}1$  and  $\text{C}3$ . The relative configurations of the phenyl rings  $\text{C}4-\text{C}9$  and  $\text{C}10-\text{C}15$  with respect to the oxetane-oxazole system are determined by their attached atoms  $\text{C}3$  ( $sp^3$ ) and  $\text{C}2$  ( $sp^2$ ), respectively. The two phenyl rings make dihedral angles of  $68.1(1)$  and  $22.4(1)^\circ$ , respectively, with the oxazole ring.

The phenanthrene is out of planarity with its dihydrobenzene ring ( $\text{C}16-\text{C}18/\text{C}23/\text{C}24/\text{C}29$ ) having a half-chair conformation. Atoms  $\text{C}16$  and  $\text{C}17$  deviate by  $0.209(2)$  and  $0.374(2)\text{ \AA}$  in opposite directions from the  $\text{C}18/\text{C}23/\text{C}24/\text{C}29$  plane.

## Experimental

The title compound was prepared by photolysis of phenanthrenequinone (0.05 *M*) in the presence of an excess amount of 2,5-

Received 11 April 2003

Accepted 17 April 2003

Online 30 April 2003

diphenyloxazole in acetonitrile and was separated by column chromatography on silica gel, with petroleum ether–ethyl acetate as eluants. Single crystals suitable for X-ray diffraction measurement were obtained by slow evaporation of a petroleum ether–acetone solution.

#### Crystal data

$C_{29}H_{19}NO_3$	$Z = 2$
$M_r = 429.45$	$D_x = 1.324 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 8.3475 (4) \text{ \AA}$	Cell parameters from 4070 reflections
$b = 11.8671 (6) \text{ \AA}$	$\theta = 2.6\text{--}28.3^\circ$
$c = 11.9106 (6) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$\alpha = 74.148 (1)^\circ$	$T = 293 (2) \text{ K}$
$\beta = 89.736 (1)^\circ$	Block, yellow
$\gamma = 72.344 (1)^\circ$	$0.50 \times 0.40 \times 0.30 \text{ mm}$
$V = 1077.59 (9) \text{ \AA}^3$	

#### Data collection

Siemens SMART CCD area-detector diffractometer	3482 reflections with $I > 2\sigma(I)$
$\omega$ scans	$R_{\text{int}} = 0.016$
Absorption correction: none	$\theta_{\text{max}} = 26.0^\circ$
5771 measured reflections	$h = -10 \rightarrow 7$
4110 independent reflections	$k = -14 \rightarrow 14$
	$l = -13 \rightarrow 14$

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0554P)^2 + 0.1949P]$
$R[F^2 > 2\sigma(F^2)] = 0.042$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.117$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.08$	$\Delta\rho_{\text{max}} = 0.20 \text{ e \AA}^{-3}$
4110 reflections	$\Delta\rho_{\text{min}} = -0.19 \text{ e \AA}^{-3}$
374 parameters	
All H-atom parameters refined	

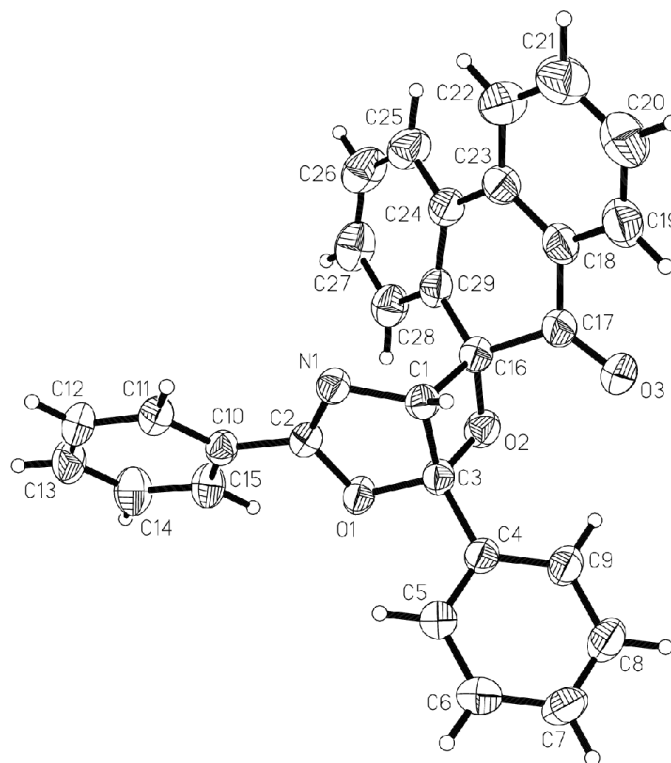
**Table 1**

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

O1–C2	1.376 (2)	N1–C1	1.449 (2)
O1–C3	1.440 (2)	C1–C3	1.543 (2)
O2–C3	1.427 (2)	C1–C16	1.586 (2)
O2–C16	1.453 (2)	C16–C29	1.506 (2)
N1–C2	1.273 (2)	C16–C17	1.519 (2)
C16–C1–C3–O2	–2.0 (1)	N1–C1–C16–C29	13.38 (18)
N1–C1–C3–O1	0.8 (1)		

All H atoms were located from a difference Fourier map and refined isotropically [ $C-H = 0.92 (2)\text{--}1.02 (2) \text{ \AA}$ ]. Owing to the large fraction of weak data at higher angles, the  $2\theta$  maximum was limited to  $52^\circ$ . The maximum and minimum electron-density peaks are located  $0.98 \text{ \AA}$  from O3 and  $1.29 \text{ \AA}$  from C6, respectively.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT* and *SADABS* (Sheldrick, 1996); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publi-



**Figure 1**

The structure of compound (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.

cation: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

The authors thank the Malaysian Government and Universiti Sains Malaysia for research grant R&D No. 304/PFIZIK/670011. AU thanks Universiti Sains Malaysia for a Visiting Postdoctoral Fellowship.

#### References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXTL*. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Usman, A., Fun, H.-K., Wang, L., Zhang, Y. & Xu, J.-H. (2003). *Acta Cryst.* **C59**. In the press.
- Zhang, Y., Wang, L. & Xu, J.-H. (2003). In preparation.